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Class IONIC AND METALLIC BONDING
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For each element below, state (i) the number of valence electrons in the atom, (ii) the electron dot structure, and (iii) the chemical symbol(s) for the most stable ion. a. Ba 2.

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Core Teaching Resources Chapters 19 Answer Key

182 Core Teaching Resources ____ 12. The molecular structure of carbon dioxide is one carbon atom with two oxygen atoms on opposite sides of it. ____ 13. Covalent bonds exist when combining atoms give up or accept electrons. ____ 14. A molecule contains two atoms. Part C Matching Match each description in Column B to the correct term in Column A.

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181 MOLECULAR COMPOUNDS 8

Core Teaching Resources Name Date
Class IONIC AND METALLIC BONDING
Practice Problems In your notebook,
answer the following SECTION 71 IONS
For each element below, state (i) the
number of valence electrons in the
atom, (ii) the electron dot structure, and
(iii) the chemical symbol(s) for the most

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Write your answer in the space provided. Date Class 11. List the five traditional branches of chemistry and explain what each involves. 12. Discuss the role of chemistry in one of these areas: Materials, Energy, Medicine and Biotechnology, Agriculture, the Environment, or the Universe. 13. looks unhealthy and that, perhaps, the grass is dying.

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