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## **Periodic Classification Of Elements Tiwari**

- Elements in a period  
have same number of  
shells. - Each period  
marks a new electronic  
shell getting filled. -  
No. of elements placed

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in a particular period  
depends upon the fact  
that how electrons are  
filled into various  
shells. [www.tiwariacademy.com](http://www.tiwariacademy.com)

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Chapter 5 Periodic  
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elements intext  
questions on Page 81  
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Video Timeline:  
0:00:00 Starts 0:08:17  
Introduction 0:11:34  
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Periodic Table- 1)  
Dobereiner's Triads  
0:19:40 2) Newland's  
Law of Octaves 0:26...

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SUPER REVISION ...**

Newland had also adjusted two elements into one slot and paired together some very unlike elements in order to make the classification work.

Ultimately this

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classification also failed  
to leave a lasting  
impression.

Mendeleev's Periodic  
Table. Now, this was  
the most successful  
attempt among all the  
earlier attempts at  
classification of  
Elements.

## **Earlier Attempts at Classification of Elements:**

### **Mendeleev ...**

In this lec we will study  
about Atomic size,



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types of radius, trend of Atomic size in periodic table down the group and across the period, radii of hydrated ...

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Atomic size ...**

In this class, Ranjeet Shahi will discuss the Periodic Classification of elements. This session would be helpful for the

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aspirants preparing for  
NEET-UG exam.

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the notes will be  
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State modern periodic  
law of classification of  
elements. (2012 D,

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2013 OD) Answer:

Modern periodic law states "that the properties of elements are a periodic function of their atomic numbers". Question 6. The atomic numbers of three elements X, Y and Z are 3, 11 and 17 respectively.

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Elements Class 10  
Important ...**

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of Elements. Shiva is  
post par aa Share with

your friends. Share 0.

Hmm... Bolo. 0 ; View

Full Answer Ye tu hai

kya jo dp lagayi hai. 0 ;

Hmm.... 0 ; Ha tu hi ko.

0 Hau mai hi hu re. 0 ...

**Shiva is post par aa -  
Chemistry - Periodic  
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CHAPTER - 5 PERIODIC

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ELEMENTS Class :- X

Subject :- Science

Name of Teacher :- Mr.

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Satish Kumar (PGT

Chem) School :- KV RRL  
Jorhat, Assam

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Mineral - Mineral -

Classification of

minerals: Since the

middle of the 19th

century, minerals have

been classified on the

basis of their chemical

composition. Under this

scheme, they are

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divided into classes according to their dominant anion or anionic group (e.g., halides, oxides, and sulfides). Several reasons justify use of this criterion as the distinguishing factor at the highest level ...

## **Mineral - Classification of minerals | Britannica**

The periodic table is basically a table in which elements are

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Elements That

Academy  
arranged on the basis  
of atomic number. A  
vertical column has  
various elements

having similar chemical  
properties. Periodic  
table groups are these  
vertical columns.

History of the Periodic  
Table. Let us start with  
the very first theory of  
periodic table. 1)

Doberiner's Triads

**Historical  
Development of  
Periodic Table:**

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**Mendeleev ...**

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## **Classification of metals pdf**

The first ten elements in modern periodic table are hydrogen, helium, lithium, beryllium, boron, carbon, nitrogen, oxygen, fluorine and neon. Out of these,



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lithium, beryllium and boron are metals, because they have 1, 2 and 3 electrons respectively in their outermost shells.

Question 6

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...

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(in Tamil) 7m 59s.

Periodic Classifications  
of Elements (Part-1)-

(in Tamil) ,, Shubham

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Classification of  
Elements and  
Periodicity in Properties  
Class 11 Notes

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Chemistry Chapter 3 •

Genesis of Periodic

Classification

Dobereiner's Triads In

1829, Dobereiner

arranged certain

elements with similar

properties in groups of

three in such a way

that the atomic mass

of the middle element

was nearly the same as

the average atomic

masses of [...]

**Classification of**

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Properties ...**

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Make social videos in  
an instant: use custom  
templates to tell the  
right story for your  
business.

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on Vimeo**

Elements, when  
*Page 20/24*

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Elements

arranged in order of increasing atomic number, lead us to the classification known as the Modern Periodic Table (Table 5.6).

Prediction of properties of elements could be made with more precision when elements were

arranged on the basis of increasing atomic number. Activity 5.2

Consider the isotopes of chlorine, Cl-35 and Cl-37

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## **CHAPTER5 Periodic Classification of Elements**

The periodic table, also known as the periodic table of elements, is a tabular display of the chemical elements, which are arranged by atomic number, electron configuration, and recurring chemical properties. The structure of the table shows periodic trends. The seven rows

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of the table, called  
periods, generally have  
metals on the left and  
nonmetals on the right.

## **Periodic table - Wikipedia**

Elements of the same  
period have the same  
number of electron  
shells. Classification of  
elements: Group 1  
contains alkali metals  
(Li, Na, K, Rb, Cs and  
Fr). The alkaline earth  
metals are metallic  
elements found in the

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group 2 of the periodic table. Elements present in groups 3 to 12 in the middle of the periodic table are called transition elements.

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